

CHEMICAL EQUATIONS

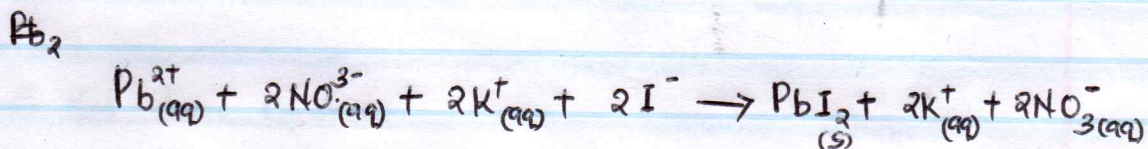
1. Reaction that produced the yellow solid.

a) Balanced Molecular Equation

Potassium iodide and Lead nitrate



b) Complete Ionic Equation



d) Net Ionic Equation



2 Reaction that produced the bubbles
Potassium carbonate and Nitric Acid

a) Balanced Molecular Equation.



b) Complete Ionic Equation.



c) Net Ionic Equation.



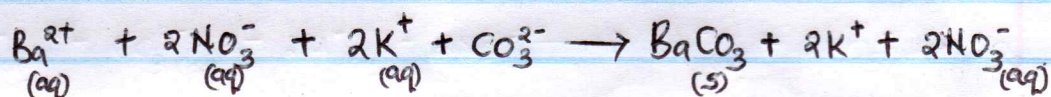
3. Reaction that produced one of the white solids

a) Balanced Molecular Equation

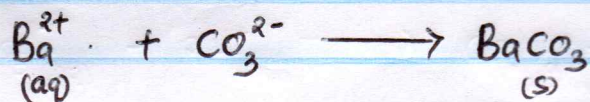
Barium Nitrate and Potassium Carbonate



b) Complete Ionic Equation



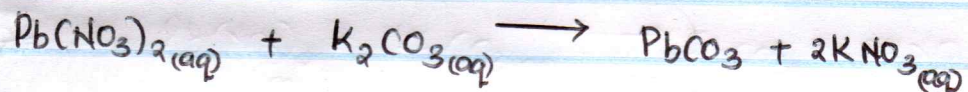
c) Net Ionic Equation



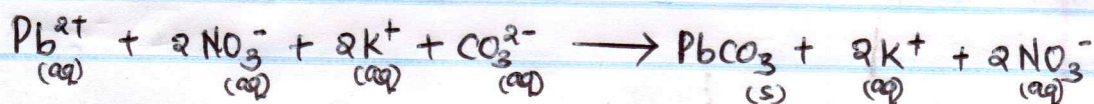
4. Reaction that produced the other one of the white solids

Lead Nitrate and Potassium carbonate

a) Balanced Molecular Equation



b) Complete Ionic equation



c) Net Ionic Equation

